

PATENT

Docket No. 0769-4582 US1

Express Mail Label No.

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

UTILITY APPLICATION AND FEE TRANSMITTAL

ASSISTANT COMMISSIONER FOR PATENTS
Box Patent Application
Washington, D.C. 20231

Sir:

Transmitted herewith for filing is the patent application of

Named Inventor(s)

(with address): Bamdad BAHAR - Apt. 25E Leatherwood Place, Baltimore, MD 21237
Alex R. HOBSON - 17 Hatters Court, Elkton, MD 21921
Jeffrey A. KOLDE - 2675 Brentwood Drive, Coopersburg, PA 18036

For: AN ULTRA-THIN INTEGRAL COMPOSITE MEMBRANE

Enclosed are:

☒ 19 page(s) of specification, 1 page(s) of Abstract, 4 page(s) of claims

☒ 2 sheets of drawing ☒ formal ☐ informal

☒ 3 page(s) of Declaration and Power of Attorney

☐ Unsigned

☐ Newly Executed

☒ Copy from prior application

☐ Deletion of inventors including Signed Statement under 37 C.F.R. § 1.63(d)(2)

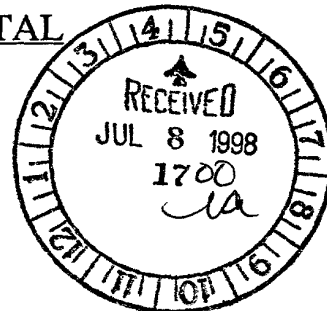
☒ Incorporation by Reference: The entire disclosure of the prior application, from which a copy of the combined declaration and power of attorney is supplied herein, is considered as being part of the disclosure of the accompanying application and is incorporated herein by reference.

☐ Microfiche Computer Program (Appendix)

☐ page(s) of Sequence Listing

☐ computer readable disk containing Sequence Listing

☐ Statement under 37 C.F.R. § 1.821(f) that computer and paper copies of the Sequence Listing are the same



Patent
Jc549 U.S. PTO
09/209932



09209932, 070898

- ☒ Assignment Papers (assignment cover sheet and assignment documents)
- ☐ A check in the amount of \$40.00 for recording the Assignment.
- ☒ Assignment papers filed in parent application Serial No. 08/903,844.
- ☐ Certification of chain of title pursuant to 37 C.F.R. § 3.73(b).
- ☐ Certified copy of Priority Document(s)
- ☐ English translation documents
- ☒ Information Disclosure Statement
- ☒ Copy of Form PTO-1449
- ☐ Copy of ___ cited references
- ☐ Preliminary Amendment
- ☐ Return receipt postcard (MPEP 503)
- ☒ This is a ☒ continuation ☐ divisional ☐ continuation-in-part (C-I-P) of prior application serial no. 08/903,844.
- ☐ Cancel in this application original claims _____ of the parent application before calculating the filing fee. (At least one original independent claim must be retained for filing purposes.)
- ☐ A Preliminary Amendment is enclosed. (Claims added by this Amendment have been properly numbered consecutively beginning with the number following the highest original claim in the prior application.
- ☒ The status of the parent application is as follows:
- ☒ A Petition For Extension of Time and a Fee therefor has been or is being filed in the parent application to extend the term for action in the parent application until July 8, 1998.
- ☒ A copy of the Petition for Extension of Time in the co-pending parent application is attached.
- ☐ No Petition For Extension of Time and Fee therefor are necessary in the co-pending parent application.
- ☐ Please abandon the parent application at a time while the parent application is pending or at a time when the petition for extension of time in that application is granted and while this application is pending has been granted a filing date, so as to make this application co-pending.
- ☐ Transfer the drawing(s) from the parent application to this application.
- ☒ Amend the specification by inserting before the first line the sentence:
This is a ☒ continuation ☐ divisional ☐ continuation-in-part of co-pending application Serial No. 08/903,844 filed July 31, 1997, which is a continuation of application Serial No. 08/339,425, filed November 14, 1994, now abandoned.

I. CALCULATION OF APPLICATION FEE

I. CALCULATION OF APPLICATION FEE						Basic Fee
	Number Filed		Number Extra		Rate	\$ 790.00
Total Claims	26	-20=	6	x	\$22.00	\$132.00
Independent Claims	4	- 3=	1	x	\$82.00	\$82.00
Multiple Dependent Claims						
	[] yes		Additional fee	=	\$270.00	\$
	[] no		Additional fee	=	NONE	

Total: \$1,004.00

- ☐ A statement claiming small entity status is attached or has been filed in the above-identified parent application and its benefit under 37 C.F.R. § 1.28(a) is hereby claimed. Reduced fees under 37 C.F.R. § 1.9(F) (50% of total) paid herewith \$_____.
- ☐ A check in the amount of \$_____ in payment of the application filing fees is attached.
- ☒ Charge Fee(s) to Deposit Account No. 13-4500. Order No. 0769-4582. A DUPLICATE COPY OF THIS SHEET IS ATTACHED.
- ☒ The Assistant Commissioner is hereby authorized to charge any additional fees which may be required for filing this application, or credit any overpayment to Deposit Account No. 13-4500. A DUPLICATE COPY OF THIS SHEET IS ATTACHED.

Respectfully submitted,

MORGAN & FINNEGAN, L.L.P.

By: Michael S. Marcus
Registration No. 31,727
(202) 857-7887 Telephone
(202) 857-7929 Facsimile

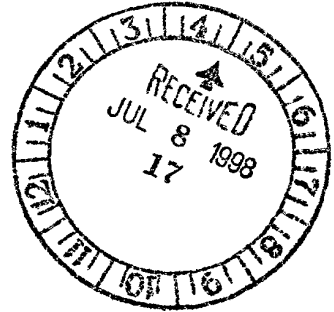
Dated: July 8, 1998

CORRESPONDENCE ADDRESS:

Morgan & Finnegan L.L.P.
345 Park Avenue
New York, New York 10154
(212) 758-4800
(202) 758-6849 Facsimile

FORM: UTL-TRAN.NY
Rev. 11/24/97

Attorney Docket No.: GT/51



UNITED STATES PATENT APPLICATION

OF

BAMDAD BAHAR, ALEX R. HOBSON AND JEFFREY A. KOLDE

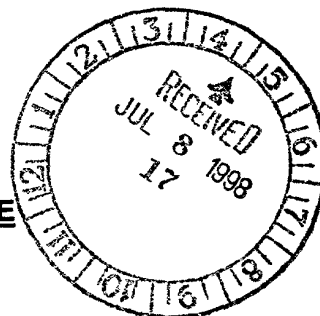
FOR

AN ULTRA-THIN INTEGRAL COMPOSITE MEMBRANE

858020" 2E550260

TITLE OF THE INVENTION

AN ULTRA-THIN INTEGRAL COMPOSITE MEMBRANE



5

FIELD OF THE INVENTION

An integral composite membrane having a thickness of less than about 1 mil (0.025 mm) is provided which is useful in electrolytic processes and other chemical separations.

10

BACKGROUND OF THE INVENTION

Ion exchange membranes (IEM) are used in fuel cells as solid electrolytes. A membrane is located between the cathode and anode and transports protons formed near the catalyst at the hydrogen electrode to the oxygen electrode thereby allowing a current to be drawn from the cell. These membranes are particularly advantageous as they replace heated acidic liquid electrolytes such as phosphoric acid fuel cells which are very hazardous.

15

Ion exchange membranes are used in chloralkali applications to separate brine mixtures and form chlorine gas and sodium hydroxide. The membrane selectively transports the sodium ions across the membrane while rejecting the chloride ions. IEM's are also useful in the area of diffusion dialysis where for example, caustic solutions are stripped of their impurities. The membranes are also useful for pervaporation and vapor permeation separations due to their ability to transfer polar species at a faster rate than their ability to transfer non-polar species.

20

25

These membranes must have sufficient strength to be useful in their various applications. Often this need for increased strength requires the membranes to be made thicker which decreases their ionic conductivity. For example, ion exchange membranes that are not reinforced such as those commercially available from E. I. DuPont de Nemours, Inc. and sold under the trademark Nafion are inherently weak at small thicknesses (e.g., less than 0.050 mm) and must be reinforced with additional materials causing the final product to have increased thickness. Moreover, these materials cannot be reliably manufactured pinhole free.

30

35

U.S. Patent No. 3,692,569 to Grot relates to the use of a coating of a copolymer of fluorinated ethylene and a sulfonyl-containing fluorinated vinyl monomer on a fluorocarbon polymer that was previously non-wettable. The fluorocarbon polymer may include tetrafluoroethylene polymers (not porous expanded PTFE). This coating provides a topical treatment to the surface so

as to decrease the surface tension of the fluorocarbon polymer. U.S. Patent No. 4,453,991 to Grot relates to a process for making a liquid composition of a perfluorinated polymer having sulfonic acid or sulfonate groups in a liquid medium that is contacted with a mixture of water and a second liquid such as a lower alcohol. The liquid made by the process may be used as a coating, a cast film, and as a repair for perfluorinated ion exchange films and membranes. Cast or coated products made with the liquid composition had thicknesses on the order of 5 mils (0.125 mm).

U.S. Patent No. 4,902,308 to Mallouk, et al. relates to a film of porous expanded PTFE having surfaces, both exterior and internal surfaces adjacent to pores, coated with a metal salt of perfluoro-cation exchange polymer. The base film of porous, expanded PTFE had a thickness of between 1 mil and 6 mils (0.025-0.150 mm). The final composite product had a thickness of at least 1 mil (0.025 mm) and preferably had a thickness of between 1.7 and 3 mils (0.043-0.075 mm). The composite product was permeable to air and the air flow as measured by the Gurley densometer ASTM D726-58 was found to be between 12 and 22 seconds.

U.S. Patent 4,954,388 to Mallouk, et al. relates to an abrasion-resistant, tear resistant, multi-layer composite membrane having a film of continuous perfluoro ion exchange polymer attached to a reinforcing fabric by means of an interlayer of porous expanded PTFE. A coating of a perfluoro ion exchange resin was present on at least a portion of the internal and external surfaces of the fabric and porous ePTFE. The composite membrane made in accordance with the teachings of this patent resulted in thicknesses of greater than 1 mil (0.025 mm) even when the interlayer of porous ePTFE had a thickness of less than 1 mil (0.025 mm).

U.S. Patent 5,082,472 to Mallouk, et al. relates to a composite membrane of microporous film in laminar contact with a continuous perfluoro ion exchange resin layer wherein both layers have similar area dimensions. Surfaces of internal pores of ePTFE may be coated at least in part with perfluoro ion exchange resin coating. The membrane of ePTFE had a thickness of about 2 mils (0.050 mm) or less and the perfluoro ion exchange layer in its original state had a thickness of about 1 mil (0.025 mm). The ePTFE layer of this composite membrane imparted mechanical strength to the composite structure and the pores of the ePTFE were preferably essentially unfilled so as to not block the flow of fluids.

U.S. Patents 5,094,895 and 5,183,545 to Branca, et al. relate to a composite porous liquid-permeable article having multiple layers of porous

5 U.S. Patent No. 4,341,615 to Bachot, et al. relates to a fluorinated resin
base material treated with a copolymer of an unsaturated carboxylic acid and
a non-ionic unsaturated monomer for use as a porous diaphragm in the
electrolysis of alkaline metal chlorides. The fluorinated resin base material
may be reinforced with fibers such as asbestos, glass, quartz, zirconia,
0 carbon, polypropylene, polyethylene, and fluorinated polyhalovinylidene (col.
2, lines 13-17). Only 0.1 to 6 percent of the total pore volume of the support
sheet is occupied by the carboxylic copolymer.

20 U.S. Patent 4,865,925 to Ludwig, et al. relates to a gas permeable electrode for electrochemical systems. The electrode includes a membrane located between and in contact with an anode and a cathode. The membrane, which may be made of expanded polytetrafluoroethylene, may be treated with an ion exchange membrane material with the resulting membrane
25 maintaining its permeability to gas. Membrane thicknesses are described to be between 1 and 10 mils, (0.025-0.25 mm), with thicknesses of less than 5 mils (0.125) to be desirable. Examples show that membrane thicknesses range from 15 to 21 mils.

Japanese Application No. 62-280230 and 62-280231 relate to a composite structure in which a scrim or open fabric is heat laminated and

encapsulated between a continuous perfluoro ion exchange membrane and an ePTFE sheet thus imparting tear strength to the structure. The composite membrane was not used for ionic conduction.

Additional research has also been conducted on the use of perfluorosulfonic acid polymers with membranes of expanded porous polytetrafluoroethylene such as that described in Journal Electrochem. Soc., Vol. 132, No. 2, February 1985, p. 514-515. The perfluoro ion exchange material was in an ethanol based solvent without the presence of water or surfactant. Moreover, ultrasonic energy in the treatment of this membrane.

Heretofore and as represented by the references discussed above, there is a need for an integral ultra-thin strong ion exchange composite membrane, with long term chemical and mechanical stability that has a thickness before swelling of at most 1 mil (0.025 mm), with more than 90% of the pore volume of the membrane filled with a perfluoro ionomer to render it at least substantially occlusive and that is capable of swelling without deterioration of mechanical properties.

SUMMARY OF THE INVENTION

An ultra-thin integral composite membrane is provided including a porous polymeric membrane having a structure of micropores of polymer with a porosity of greater than 35%, an average pore diameter of less than 10 microns and a thickness of at most 0.025 mm and a perfluoro ion exchange polymer impregnated within the micropores so as to render the micropores substantially occlusive, wherein the composite membrane is impermeable to gases and liquids and is substantially free of pinholes. Porous polymeric membranes suitable for this invention include membranes made of perfluoroalkyloxy resin, fluorinated ethylene propylene, polyolefins, polyamides, cellulose, polycarbonates, fluorinated and chlorinated polymers, and polysulfones. Perfluoro ion exchange materials suitable for use with this invention include perfluorinated sulfonic acid resin, perfluorinated carboxylic acid resin, polyvinyl alcohol, divinyl benzene, and styrene based polymers. A reinforcement backing may also be provided.

Methods for making the ultra-thin integral composite membranes are also provided.

BRIEF DESCRIPTION OF THE DRAWINGS

5 Figure 1 is a schematic cross-section of the composite membrane that is fully impregnated with the ion exchange material.

Figure 2 is a schematic cross-section of the composite membrane that is fully impregnated with the ion exchange material and has a backing material attached thereto.

10 Figure 3 is a photomicrograph of a cross-section of expanded PTFE that has not been treated with any ion exchange material at a magnification of 2.5 kX.

Figure 4 is a photomicrograph of a cross-section of expanded PTFE fully impregnated with an ion exchange material at a magnification of 5.1 kX.

DETAILED DESCRIPTION OF THE INVENTION

15 An ultra-thin composite membrane is provided and includes a base material of microporous membrane with a thickness less than 1 mil (0.025 mm) having a microstructure of micropores and perfluoro ion exchange resin that
20 substantially impregnates the microporous membrane so as to occlude the micropores. The ultra-thin composite membrane may be employed in many different types of applications including for example, chemical separation, electrolysis in fuel cells and batteries, pervaporation, gas separation, dialysis separation, industrial electrochemistry such as chlor-alkali, and other
25 electrochemical devices, catalysis as a super acid catalyst and use as a catalyst support in enzyme immobilization.

The ultra-thin composite membrane is mechanically strong and is substantially and uniformly pore occlusive so that it is particularly useful as an ion exchange material. Ultra-thin is hereby defined as 1 mil (0.025 mm) or
30 less. Uniform is hereby defined as continuous impregnation with the ion exchange material so that no pin holes or other discontinuities exist within the composite structure. In addition, pore occlusive is hereby defined as pores being substantially impregnated (i.e., at least 90%) with the perfluoro ion exchange material rendering the final material air impermeable with a Gurley
35 number of infinity.

The microporous membrane which serves as the base material for the composite has a porosity of greater than 35% and preferably has a porosity of between 70-95%. The pores of the microstructure have a diameter less than

0920932 070398 866020 2E660260

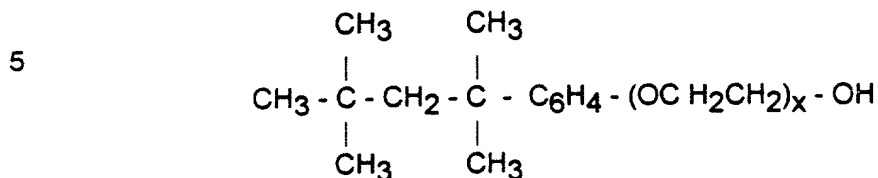
10 μm , are preferably between 0.05 and 5 μm , and are most preferably about 0.2 μm . The thickness of the membrane is at most 1 mil (0.025 mm) preferably between 0.06 mils (0.19 μm) and 0.8 mils (0.02 mm), and most preferably between 0.50 mils (0.013 mm) and 0.75 mils (0.019 mm). Materials
5 from which this microporous membrane can be made include for example, perfluoroalkyloxy (PFA), fluorinated ethylene propylene (FEP), polyolefins, polyamides, cellulose, polycarbonates, fluorinated and/or chlorinated polymers, and polysulfones. A most preferred material is expanded porous polytetrafluoroethylene (PTFE) made in accordance with the teachings of U.S.
10 Patent No. 3,593,566 herein incorporated by reference. This material is commercially available in a variety of forms from W. L. Gore & Associates, Inc., of Elkton, MD, under the trademark GORE-TEX®. The expanded PTFE membrane can be made in a number of thickness ranging from 0.00025 inches to 0.125 inches (6 μm to 3 mm) with the preferred thickness for the
15 present invention being at most 1 mil (0.025 mm) and most preferably between 0.50 mils (0.013 mm) and 0.75 mils (0.019 mm). The expanded PTFE membrane can be made with porosities ranging from 20% to 98%, with the preferred porosity for the present invention being 70-95%. Figure 3 shows a photomicrograph of the internal microstructure of expanded PTFE used as
20 the base material.

An ion exchange material dissolved in a solvent and mixed with a surfactant is uniformly applied so as to impregnate and occlude the micropores of the base material. Suitable ion exchange materials include for
25 example, perfluorinated sulfonic acid resin, perfluorinated carboxylic acid resin, polyvinyl alcohol, divinyl benzene, styrene-based polymers and metal salts with or without a polymer. A mixture of these ion exchange materials may also be employed in treating a membrane. Solvents that are suitable for use with the ion exchange material include for example, alcohols, carbonates, THF (tetrahydrofuran), water, and combinations thereof.

30 A surfactant having a molecular weight of greater than 100 must be employed with the ion exchange material to ensure impregnation of the pores. Surfactants or surface active agents having a hydrophobic portion and a hydrophilic portion may be utilized. Preferable surfactants are those having a molecular weight of greater than 100 and may be classified as anionic,
35 nonionic, or amphoteric which may be hydrocarbon or fluorocarbon-based and include for example, Merspol®, a hydrocarbon based surfactant or Zonyl®, a fluorocarbon based surfactant, both commercially available from E. I. DuPont de Nemours, Inc. of Wilmington, DE.

090993 070898
B5B020" 2E660260

A most preferred surfactant is a nonionic material, octylphenoxy polyethoxyethanol having a chemical structure:



10 where $x = 10$ (average)

known as Triton® X100, commercially available from Rohm & Haas of Philadelphia, PA.

15 Figure 1 shows a schematic view of the composite membrane with ion exchange material 2 and the base material 4 so that the micropores of the interior of the base material 4 are fully impregnated. The final composite membrane has a uniform thickness free of any discontinuities or pinholes on the surface. The micropores of the membrane are 100% occluded thus causing the composite membrane to be impermeable to liquids and gases.

20 Figure 4 shows a scanning electron photomicrograph of this composite membrane.

Alternatively, the ion exchange material and surfactant mixture 2 may be applied to the membrane 4 so that most of the pores are uniformly treated rendering the membrane substantially impregnated with the ion exchange material. The composite membrane is still free of any discontinuities or pinholes.

Optionally, and as shown schematically in Figure 2, the composite membrane may be reinforced with a woven or non-woven material 6 bonded to one side of the membrane. Suitable woven materials include for example, 30 scrim made of woven fibers of expanded porous polytetrafluoroethylene, commercially available from W. L. Gore & Associates, Inc., of Elkton, Maryland; webs made of extruded or oriented polypropylene netting commercially available from Conwed, Inc. of Minneapolis, MN; and woven materials of polypropylene and polyester of Tetko Inc., of Briarcliff Manor, New York. Suitable non-woven materials include for example, a spun-bonded polypropylene commercially available from Reemay Inc. of Old Hickory, TN.

The treated membrane may then be further processed to remove the surfactant with the use of various low molecular weight alcohols. This is accomplished by soaking or submerging the membrane in a solution of, for

868020" 2E60250

example, water, isopropyl alcohol, methanol and/or glycerin. During this step, the surfactant which was originally mixed in solution with the perfluoro ion exchange material is removed. Slight swelling of the membrane occurs. The perfluoro ion exchange material remains within the pores of the base material as it is not soluble in the lower molecular weight alcohol.

The membrane is further treated by boiling in a suitable swelling agent, preferably water causing it to then slightly swell in the x and y direction. Additional swelling occurs in the z-direction. A composite membrane results having a higher ion transport rate (at least 20 times higher than in its unswollen state) that is also strong. The swollen membrane retains its mechanical integrity unlike the membranes consisting only of the perfluoro ion exchange material and simultaneously maintains desired ionic transport characteristics. A correlation exists between the content of the swelling agent within the membrane structure and transport properties of the membrane. A swollen membrane will transport chemical species faster than an unswollen membrane.

Although the membrane has excellent long term chemical stability, it can be susceptible to poison by organics. The organics can be removed by regeneration in which the membrane is boiled in a strong acid such as nitric or chromic acid.

To prepare the inventive membrane, a support structure such as a polypropylene woven fabric may first be laminated to the untreated base membrane by any conventional technique including hot roll lamination, ultrasonic lamination, adhesive lamination, forced hot air lamination so long as the technique does not damage the integrity of the membrane. A solution is prepared containing a perfluorosulfonic acid resin in solvent mixed with one or more surfactants. The solution may be applied to the membrane by any conventional coating technique including forwarding roll coating, reverse roll coating, gravure coating, doctor coating, kiss coating, as well as dipping, brushing, painting, and spraying so long as the liquid solution is able to penetrate the interstices and micropores of the base material. Excess solution from the surface of the membrane may be removed. The treated membrane is then immediately placed into an oven to dry. Oven temperatures may range from 60-200°C, preferably 120-160°C so as to lock the perfluoro ion polymer inside the membrane and prevent it from migrating to the surface during drying. This step may be repeated until the membrane becomes completely transparent. Typically between 2 to 8 treatments are required but the actual number of treatments is dependent on the surfactant concentration and

thickness of the membrane. If the membrane is prepared without a support structure, both sides of the membrane may be treated simultaneously thereby reducing the number of treatments required.

5 The oven treated membrane is then soaked in a low molecular weight alcohol, as described above to remove the surfactant. The membrane is then boiled as described above in a swelling agent under pressure ranging from 0 to 20 atmospheres absolute thereby increasing the amount of swelling agent the treated membrane is capable of holding.

10 Alternatively, the ion exchange material may be applied to the membrane without the use of a surfactant. This procedure requires additional exposure to the perfluoro ion exchange resin but does not then need to be soaked in alcohol. A vacuum may also be used to draw the ion exchange material into the membrane.

15 Another alternative to the process of preparing the inventive membrane involves the selection and use of a surfactant having low water solubility with the perfluoro ion solution. Surfactants with low water solubility include Zonyl® FSO, a fluorocarbon based surfactant commercially available from E. I. DuPont de Nemours, Inc. By using this type of surfactant, the heat treatment step may be eliminated. The resulting ion exchange treated membrane made
20 by this process may be used for different aqueous applications and other chemical environments without any effect due to the surfactant.

25 Because the base membrane is exceptionally thin (at most 1 mil) (0.025 mm) with the resulting composite membrane being very thin and only marginally distorted in the x and y directions, it is able to selectively transport ions at a faster rate than heretofore has been achieved with only a slight lowering of the selectivity characteristics of the membrane.

The following testing procedures were employed on the samples prepared as described in the examples described below.

30

TEST PROCEDURES

AIR PERMEABILITY - Gurley Number Method

35 The resistance of samples to air flow was measured by a Gurley densometer (ASTM D726-58) manufactured by W. & L. E. Gurley & Sons. The results were reported in terms of Gurley Number defined as the time in seconds for 100 cubic centimeters of air to pass through 1 square inch (6.45 sq. cm.) of a test sample at a pressure drop of 4.88 inches (12.4 cm.) of water.

STRENGTH MODULUS

Strength testing was carried out on an Instron Model 1122. Samples were one inch wide. Gauge length (distance between clamps) was two inches (5.08 cm.). Samples were pulled at a rate of 500% per minute. The cross head speed was 20 inches per minute.

THICKNESS

Thickness of the base material was determined with the use of a snap gauge (Johannes Kafer Co. Model No. F1000/302). Measurements were taken in at least four areas of each specimen. Thickness of the dried composite membrane were also obtained with the use of the snap gauge. Thicknesses of swollen samples were not measurable due to the compression or residual water on the surface of the swollen membrane with the snap gauge. Thickness measurements of the swollen membrane were also not able to be obtained with the use of scanning electron microscopy due to interferences with the swelling agents.

MOISTURE VAPOR TRANSMISSION RATE (MVTR)

Potassium Acetate Method

Moisture vapor transmission rates were determined by the following procedure. Approximately 70 ml. of a solution consisting of 35 parts by weight of potassium acetate and 15 parts by weight of distilled water was placed into a 133 ml. polypropylene cup, having an inside diameter of 6.5 cm. at its mouth. An expanded polytetrafluoroethylene (PTFE) membrane having a minimum MVTR of approximately $85,000 \text{ g/m}^2\text{-24 hr}$ as tested by the method described in U.S. Patent 4,862,730 to Crosby and available from W. L. Gore & Associates, Inc. of Newark, Delaware, was heat sealed to the lip of the cup to create a taut, leakproof, microporous barrier containing the solution.

A similar expanded PTFE membrane was mounted to the surface of a water bath. The water bath assembly was controlled at 23°C plus or minus 0.2°C , utilizing a temperature controlled room and a water circulating bath.

The sample to be tested was allowed to condition at a temperature of 23°C and a relative humidity of 50% prior to performing the test procedure. Samples were placed so the microporous polymeric membrane was in contact with the expanded polytetrafluoroethylene membrane mounted to the surface of the water bath and allowed to equilibrate for at least 15 minutes prior to the introduction of the cup assembly.

The cup assembly was weighed to the nearest 1/1000g. and was placed in an inverted manner onto the center of the test sample.

Water transport was provided by the driving force between the water in the water bath and the saturated salt solution providing water flux by diffusion in that direction. The sample was tested for 10 minutes and the cup assembly was then removed, weighed again within 1/1000g.

The MVTR of the sample was calculated from the weight gain of the cup assembly and was expressed in grams of water per square meter of sample surface area per 24 hours.

PEEL STRENGTH

Peel strength or membrane adhesion strength tests were conducted on samples prepared with reinforced backings. Test samples were prepared having dimensions of 3 inch by 3.5 inch (7.62 cm x 8.89 cm). A 4 inch by 4 inch (10.2 cm x 10.2 cm) chrome steel plate with an Instron tensile test machine Model No. 1000 were also used. Double coated vinyl tape (3M - #419) 1 inch (2.54 cm) wide was placed over the edges of the chrome plate so that tape covered all edges of the plate.

The sample material was then mounted on top of the adhesive exposed side of the tape and pressure was applied so that sample was secured.

The plate and sample were then installed in the Instron in a horizontal position. The upper crosshead was lowered so that the jaws of the machine closed flat and tightened upon the sample. The upper crosshead was then slowly raised pulling the layer of perfluoro ion material from the reinforced backing. When the composite membrane had been detached from the reinforced backing, the test was complete. Adhesion strength was estimated from the average strength needed to pull the composite membrane from the reinforced backing.

ELECTRICAL CONDUCTANCE

The electrical conductance in the Z-direction (otherwise known as through conductance) was measured. A sample of swollen composite membrane (cut to a 1 inch diameter circle) was placed between two 0.680 inch (1.73 cm) diameter copper contacts. A 5 lb. (2268 g) weight was placed above the top contact. The contacts were connected to a Hewlett Packard Model 3478A multimeter. The resistance was then read. Prior to this measurement, the thickness of the dried preswollen composite membrane was

5

wherein R = resistance measured in ohms
C = conductance measured in mhos

10

15

25

30

The percent weight change on samples was also prepared. Here composite membranes were weighed prior to boiling and swelling and then after swelling. All weight measurements were taken with Mettler Balance, Model No. AT400. The percent weight change was then calculated.

35

Example 1:

A sample of expanded polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 3,593,566, herein incorporated by reference. The membrane, with a nominal thickness of 0.75
5 mils (0.02 mm) and a 0.2 micrometer pore size, was mounted on a 6 inch wooden embroidery hoop. A solution was prepared comprising 95% by volume of a perfluorosulfonic acid/tetra-fluoroethylene copolymer resin (in H+ form) in a solution of low molecular weight alcohols comprising propanol, butanol, and methanol known as Nafion NR-50 (1100 EW) commercially
10 available from E.I. DuPont de Nemours, Inc. and 5% of a nonionic surfactant of octyl phenoxy poly ethoxyethanol known as Triton X100 commercially available from Rohm & Haas of Philadelphia, PA. This solution was brushed on both sides of the membrane so as to impregnate and substantially occlude the micropore structure. The sample was then dried in the oven at 140°C for
15 30 seconds. The procedure was repeated two more times to fully occlude the micropores. The sample was then soaked in isopropanol for 5 minutes to remove the surfactant. After rinsing with distilled water and allowing to dry at room temperature, a final coat of the Nafion-surfactant solution as described above was applied. The wet membrane was again dried in the oven at 140°C
20 for 30 seconds and soaked in isopropanol for 2 minutes. The membrane was finally boiled in distilled water for 10 minutes under atmospheric pressure to swell the treated membrane. Gurley numbers for this material are summarized in Table 3. Ionic conductive rates are summarized in Table 4. The strength modulus may be found in Table 5; percent linear expansion may be found in
25 Table 6; and percent weight change of this sample may be found in Table 7. The swollen membrane was later dried to a dehydrolyzed state in an oven at 140°C for 30 seconds. The thickness of the dried composite membrane was measured and found to be approximately the same thickness as the base material.

Example 2

A sample of expanded porous PTFE membrane made in accordance with the teachings of U.S. Patent No. 3,593,566 having a pore size of 0.2 micrometers and nominal thickness of 0.75 mils (0.02 mm) and a Gurley
35 Densometer air flow of 2-4 seconds was placed on top of a netting of polypropylene obtained from Conwed Plastics Corp. of Minneapolis, MN. The two materials were bonded together on a laminator with 10 psig pressure, a speed of 15 feet per minute and a temperature of 200°C. No adhesives were

used. The reinforced membrane sample was then placed on a 6 inch wooden embroidery hoop. A solution of 96% by volume of a perfluorosulfonic acid TFE copolymer resin in alcohol Nafion NR-50 (1100 EW) commercially available from E.I. DuPont de Nemours, Inc. and 4% of the nonionic surfactant Triton X-100 obtained from Rohm & Haas was prepared. This solution was brushed on the membrane side only to substantially occlude the micropores and the sample was dried in an oven at 130°C. This procedure was repeated three more times to fully occlude the micropores. The sample was then baked in an oven at 140°C for 5 minutes. The sample was soaked in isopropanol for 5 minutes to remove the surfactant. The membrane was then boiled in distilled water for 30 minutes under atmospheric pressure causing the treated membrane to swell. Gurley numbers for this material are summarized in Table 3.

This sample was tested for its peel strength in accordance with the method described above. The linear bond strength was found to be 2.06 lb./sq. in. (1450 kg/m²).

Example 3

A sample of expanded porous polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 3,593,566, having a thickness of 0.5 mils (0.01 mm) with a pore size of 0.2 micrometer was mounted on a 6 inch wooden embroidery hoop. A solution of 100% Nafion resin solution, perfluorosulfonic acid/TFE copolymer resin in a solvent mixture of propanol, butanol, and methanol known commercially from E. I. DuPont de Nemours, Inc. as Nafion® solution NR-50 (1100 EW) without the addition of any surfactants was brushed onto both sides of the membrane to substantially occlude the micropores. The sample was then placed in an oven at 160°C to dry. This procedure was repeated four more times until the membrane was completely transparent and the micropores were fully occluded. The sample was then boiled in distilled water for 30 minutes at atmospheric pressure causing the membrane to swell. Gurley numbers for this material are summarized in Table 3. The electrical conductivity was measured and summarized in Table 2.

Example 4

A sample of expanded porous polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 3,593,566 having a thickness of 0.5 mils (0.01 mm) and a pore size 0.2 micrometers was mounted

on a 6 inch wooden embroidery hoop. A solution of 99% by volume Nafion NR-50 commercially available from E.I. DuPont de Nemours, Inc. and 1% surfactant mixture was prepared. The surfactant mixture consisted of 50% of a nonionic surfactant, Triton X-100 commercially available from Rohm & Haas Corp. and 50% Zonyl FSO commercially available from E.I. DuPont de Nemours, Inc. This solution was brushed on both sides of the membrane and was allowed to dry at room temperature. This procedure was repeated 4 more times until the sample was completely transparent and to fully occlude the micropores. The sample was not treated so as to remove the surfactant. The composite membrane was boiled in distilled water for 5 minutes causing the membrane to swell. The Gurley number for this material is summarized in Table 3.

Example 5

A sample of expanded porous polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 3,593,566, having a thickness of 0.5 mils (0.01 mm) with a pore size of 0.2 micrometer was mounted onto a 6 inch wooden embroidery hoop. A solution of 95% by volume Nafion NR-50 (1100 EW) commercially available from E.I. DuPont de Nemours, Inc. and 5% of a nonionic surfactant, Triton X-100 commercially available from Rohm & Haas was prepared. The solution was brushed on both sides of the membrane with a foam brush and the excess was wiped off. The wet membrane was dried in an oven at 140°C for 30 seconds. Three additional coats of solution were applied to the membrane in the same manner to fully occlude the micropores. The membrane was then soaked in isopropanol for 2 minutes to remove the surfactant. The membrane was rinsed with distilled water and allowed to dry at room temperature. A final treatment of the Nafion-Triton solution was applied. The wet membrane was dried in the oven at 140°C for 30 seconds, then soaked in isopropanol for 2 minutes. Finally, the membrane was boiled in distilled water for 5 minutes. Moisture vapor transmission rates for the treated membrane were measured and are summarized in Table 1. The Gurley number of the treated membrane are summarized in Table 3.

Example 6

A sample of expanded porous polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 53,593,566, having a nominal thickness of 0.75 mils (0.02 mm) and a pore size of 0.2 micrometers

was mounted onto a 6 inch wooden embroidery hoop. The Gurley Densometer air flow on this membrane was 2-4 seconds. A solution of 95% by volume Nafion NR-50 (1100 EW) commercially available from E.I. DuPont de Nemours, Inc. and 5% Triton X-100 non-ionic surfactant from Rohm & Haas was prepared. The solution was brushed on both sides of the membrane with a foam brush and the excess was wiped off. The wet membrane was dried in the oven at 140°C for 30 seconds. Three additional coats of solution were applied in the same manner. The membrane was then soaked in isopropanol for 2 minutes. After rinsing with distilled water and allowing to dry at room temperature, a final coat of the Nafion surfactant solution was applied. The wet membrane was dried in the oven at 140°C for 30 seconds, then soaked in isopropanol for 2 minutes. This material was not boiled. No swelling other than the minor swelling during the surfactant removal occurred. The ionic conduction rate for this material is presented in Table 4.

Example 7

A sample of expanded porous polytetrafluoroethylene membrane made in accordance with the teachings of U.S. Patent No. 3,593,566, having a nominal thickness of 0.75 mils (0.02 mm) and a pore size of 0.2 micrometers was mounted onto a 5 inch plastic embroidery hoop. The Gurley Densometer air flow on this membrane was 2-4 seconds. A solution of 95% by volume Nafion NR-50 (1100 EW) commercially available from E.I. DuPont de Nemours, Inc. and 5% Triton X-100 non-ionic surfactant from Rohm & Haas was prepared. The solution was brushed on both sides of the membrane with a foam brush and the excess was wiped off. The wet membrane was dried in the oven at 140°C for 30 seconds. Two additional coats of solution were applied in the same manner so as to fully occlude the micropores. The membrane was then soaked in isopropanol for 2 minutes. After rinsing with distilled water and allowing to dry at room temperature, a final coat of the same Nafion NR-50 Triton X-100 solution was applied. The wet membrane was dried in the oven at 140°C for 30 seconds and then soaked in isopropanol for 2 minutes to remove the surfactant. The sample was rinsed and dried at room temperature. No boiling occurred.

This sample was weighed before it was mounted on the 5 inch plastic hoop. Following treatment, it was removed from the hoop and weighed again. The ion exchange polymer content was directly calculated by determining the weight change before and after treatment. The ion exchange content for this sample was found to be 98.4 mg or 9.81 grams per square meter of

membrane. A sample of Nafion 115 (5 mils, 0.13 mm) commercially available from E. I. DuPont de Nemours, Inc. was cut to a 1 inch (25.4 mm) by 1 inch (25.4 mm) sample, weighed and found to be 216 grams per square meter.

Nafion Comparative Samples

Nafion 117, a perfluorosulfonic acid cation exchange membrane, unreinforced film of 1100 equivalent weight commercially available from E. I. DuPont de Nemours Co., Inc., having a quoted nominal thickness of 7 mils (0.18 mm) was obtained. The samples, originally in the hydrated swollen state were measured in the x- and y-directions and weighed. The samples were then dried at room temperature to an unswollen state and then remeasured from which expansion and weight change measurements found in Tables 6 and 7 were calculated. Nafion 115, a perfluorosulfonic acid cation exchange membrane, unreinforced film of 1100 equivalent weight also commercially available from E. I. DuPont de Nemours, Inc., having a nominal thickness of 5 mils (0.1 mm) was obtained. This sample was also obtained commercially in the hydrated swollen state.

TABLE 1

Moisture Vapor Transmission Rates (MVTR)

Sample ID*	MVTR (grams/m ² -24 hrs.
5	25,040
Nafion 117	23,608

*Measurements were obtained on samples in their swollen state.

TABLE 2

Electrical Conductivity

Sample ID*	Conductivity (micromhos)
3	1,277
Nafion 117	1,214

*Measurements were obtained on samples in their swollen state.

TABLE 6
Percent Linear Expansion

Sample ID	Unswollen (x) (mm)	Swollen (x) (mm)	Average % Expansion in x-direction	Unswollen (y) (mm)	Swollen (y) (mm)	Average % Expansion in y-direction
1	124.4	124.4	--	123.3	123.3	--
Nafion 117	125.5	137.7	+9.7	127.3	149.7	+17

5

TABLE 7
Percent Weight Change

Sample ID	Unswollen wt (g)	Swollen wt (wet) (g)	% weight change (wet)
1	0.2515	1.0273	+308%
Nafion 117	5.5700	7.5106	+35%

363020" 2E650260

10

- 3

- (a) a porous membrane having a microstructure of micropores of polymer with a porosity of greater than 35%, an average pore diameter of less than 10 μm and a thickness of at most 0.025 mm; and
- 5 (b) a perfluoro ion exchange material fully impregnated within the micropores of the polymeric membrane so as to render them fully occlusive, wherein the composite membrane is air impermeable with a Gurley number of infinity, and is substantially free of pinholes.
- 10 10. An ultra-thin composite membrane as described in claim 9 wherein the porous polymeric membrane is selected from the group including perfluoroalkyloxy resin, fluorinated ethylene propylene, polyolefins, polyamides, cellulotics, polycarbonates, fluorinated and chlorinated polymers, and polysulfones.
- 15 11. An ultra-thin integral composite membrane as described in Claim 9 wherein the porous polymeric membrane is expanded porous polytetrafluoroethylene.
12. An ultra-thin integral composite membrane as described in Claim 9 wherein the perfluoro ion exchange material is selected from the group including perfluorinated sulfonic acid resin, perfluorinated carboxylic acid resin, polyvinyl alcohol, divinyl benzene, and styrene-based polymers.
- 20 13. An ultra-thin integral composite membrane as described in Claim 9 wherein the perfluoro ion exchange material is a perfluorosulfonic acid/tetrafluoroethylene copolymer resin dissolved in a solvent solution selected from the group including water, ethanol, propanol, butanol, methanol and combinations thereof.
- 25 14. An ultra-thin composite membrane as described in Claim 9 further comprising a reinforcement backing bonded to the porous polymeric membrane, wherein the reinforcement backing is selected from the group including woven and nonwoven materials
- 30 15. An ultra-thin composite membrane as described in Claim 14 wherein the woven materials are selected from the group including: weaves of expanded porous polytetrafluoroethylene fibers, webs of polypropylene, and netting of polypropylene.
- 35 16. An ultra-thin composite membrane as described in Claim 14 wherein the nonwoven material is spun-bonded polypropylene.
17. A method of making an ultra-thin composite membrane comprising the steps of:

068040 2560260

- 35

24. A method of making an ultra-thin composite membrane as described in Claim 18 further comprising the step of selecting a lower molecular weight alcohol for removing the surfactant from the group including isopropanol, methanol, and glycerin.
- 5 25. A method of making an ultra-thin composite membrane as described in Claim 17 wherein the impregnating step is selected from the group including forward roll coating, reverse roll coating, gravure coating, doctor coating, kiss coating, dipping, brushing, painting and spraying.
- 10 26. A process for regenerating an ultra-thin composite membrane of a porous polymeric membrane that has been impregnated with a perfluoro ion exchange material comprising the steps of:
- (a) selecting a strong acid from the group including nitric and chromic acids; and
 - (b) boiling said composite membrane with the strong acid.

15

868040 2E60260

ABSTRACT

- 5 An integral composite membrane is provided including a porous polymeric membrane impregnated with a perfluoro ion exchange material to make the micropores of the membrane occlusive and a surfactant having a molecular weight greater than 100 wherein the thickness of the composite membrane is less than .025 mm. Methods for making and regenerating the integral composite membrane are also provided.

2025/01/20 10:20:00



FIG. 1

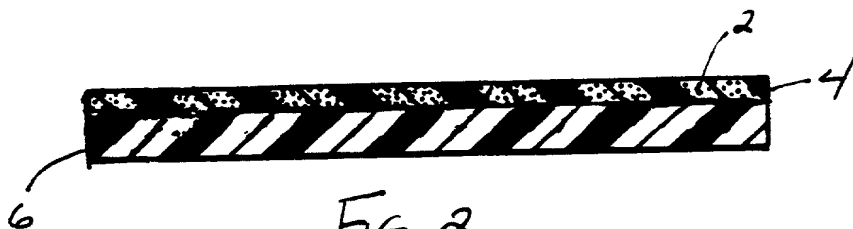


FIG. 2

0900993 070899 868020 2660260

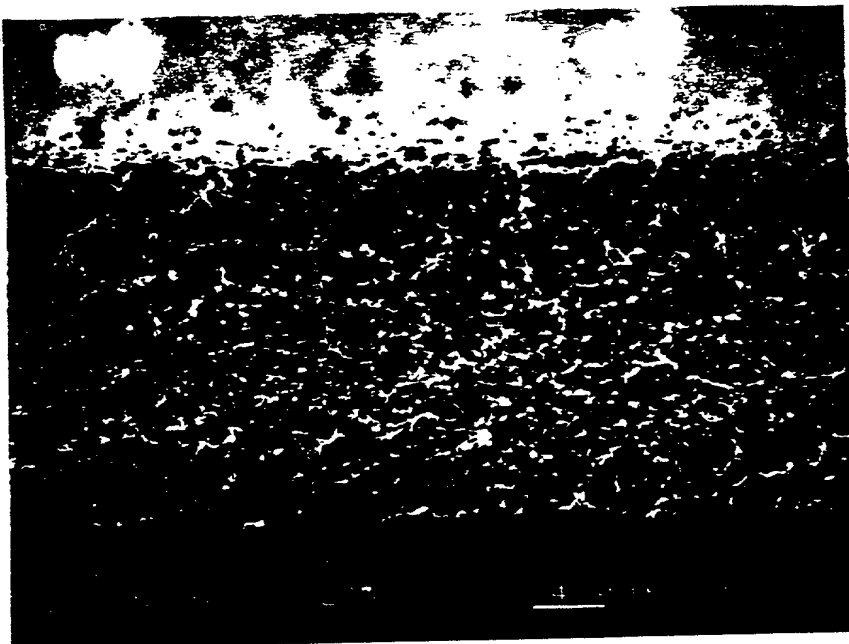


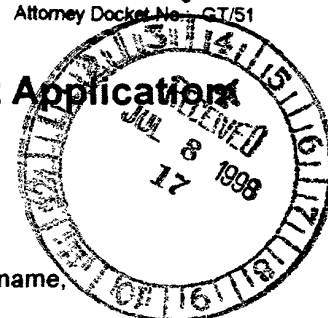
FIGURE 3



FIGURE 4

09209332 070898 868070 22660260

Declaration and Power of Attorney For Patent Application English Language Declaration



As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled

AN ULTRA-THIN INTEGRAL COMPOSITE MEMBRANE

the specification of which

(check one)

☐ is attached hereto.

☒ was filed on November 14, 1994 as

Application Serial No. 08/339,425

and was amended on December 8, 1994

(if applicable)

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code, §119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Prior Foreign Application(s)			Priority Claimed	
_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/> Yes	<input type="checkbox"/> No
_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/> Yes	<input type="checkbox"/> No
_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/> Yes	<input type="checkbox"/> No

I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)

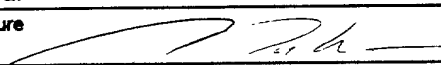
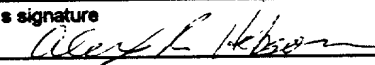
I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith. *(list name and registration number)*

Gary A. Samuels, 20,811; Dena Meyer Weker, 32,540; John S. Campbell, 28,366;
Wayne D. House, 34,623; David J. Johns, 31,527; Eric J. Sheets, 30,326

Send Correspondence to: Dena Meyer Weker, Esq., W. L. Gore & Associates, Inc.,
551 Paper Mill Road, Newark, DE 19714-9206

Direct Telephone Calls to: *(name and telephone number)*: Dena Meyer Weker, (302) 738-4880

Full name of sole or first inventor Bamdad Bahar	
Inventor's signature 	Date Dec 5 '94
Residence Apt. 25E Leatherwood Place, Baltimore, MD 21237	
Citizenship U.S.A.	
Post Office Address Apt. 25E Leatherwood Place, Baltimore, MD 21237	
Full name of second joint inventor, if any Alex R. Hobson	
Second Inventor's signature 	Date Dec 5 '94
Residence 17 Hatters Court, Elkton, MD 21921	
Citizenship U.S.A.	
Post Office Address 17 Hatters Court, Elkton, MD 21921	

(Supply similar information and signature for third and subsequent joint inventors.)

090993 070993 060200 2260260

(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)
(Application Serial No.)	(Filing Date)	(Status) (patented, pending, abandoned)

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith. *(list name and registration number)*

Gary A. Samuels, 20,811; Dena Meyer Weker, 32,540; John S. Campbell, 28,366;
Wayne D. House, 34,623; David J. Johns, 31,527; Eric J. Sheets, 30,326

Send Correspondence to: Dena Meyer Weker, Esq., W. L. Gore & Associates, Inc.,
551 Paper Mill Road, Newark, DE 19714-9206

Direct Telephone Calls to: *(name and telephone number)*: Dena Meyer Weker (302) 738-4880

Full name of third joint inventor, if any Jeffrey A. Kolde	
Third Inventor's signature <i>Jeffrey A. Kolde</i>	Date 2/7/7
Residence 2675 Brentwood Drive, Coopersburg, PA 18036	
Citizenship U.S.A.	
Post Office Address 2675 Brentwood Drive, Coopersburg, PA 18036	
Full name of fourth joint inventor, if any	
Fourth Inventor's signature	Date
Residence	
Citizenship	
Post Office Address	

(Supply similar information and signature for third and subsequent joint inventors.)

068070" 2560260